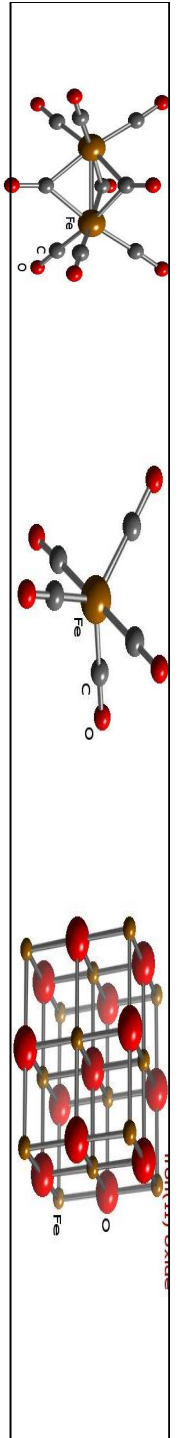


Uptake Rates as Fundamentals of Iron Availability to Phytoplankton

Yeala Shaked

Interuniversity Inst. for Marine Sciences, Eilat
Hebrew University, Jerusalem, Israel

GEOTRACES Synthesis Workshop Aug 1-4th 2016



What determines the bioavailability of iron to phytoplankton ?

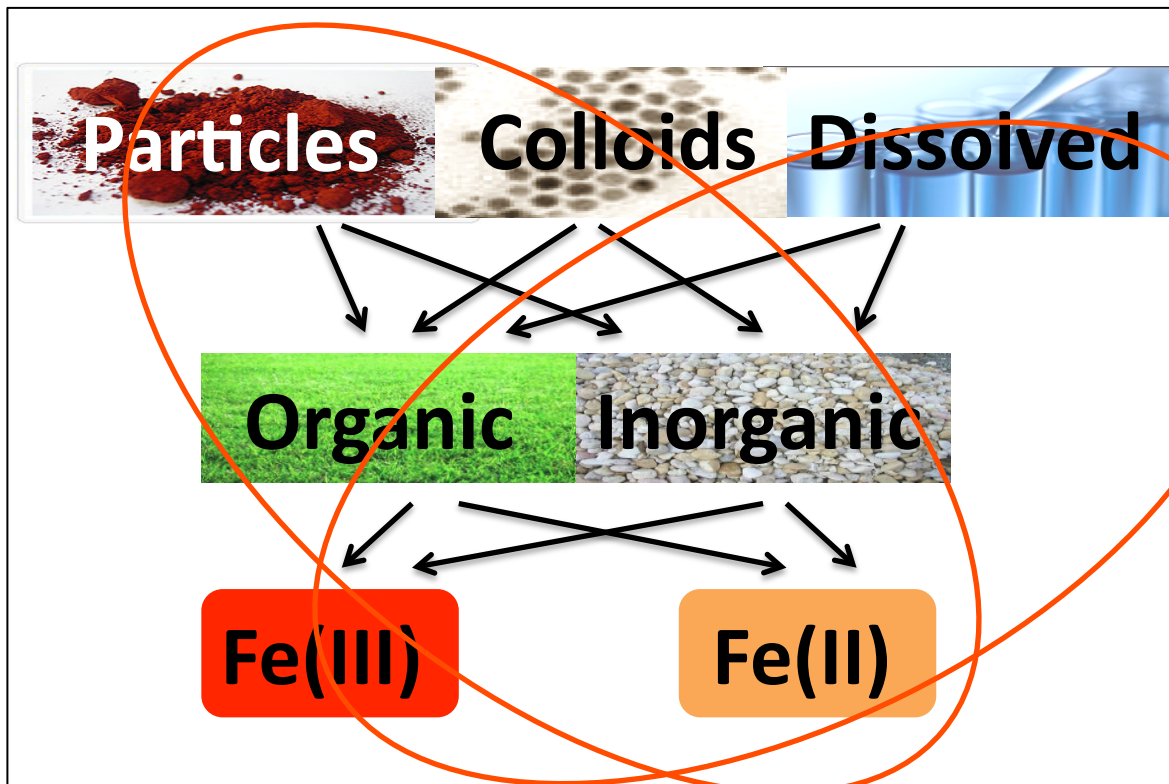
Fundamentals of Fe availability

Interactions and ecosystem processes

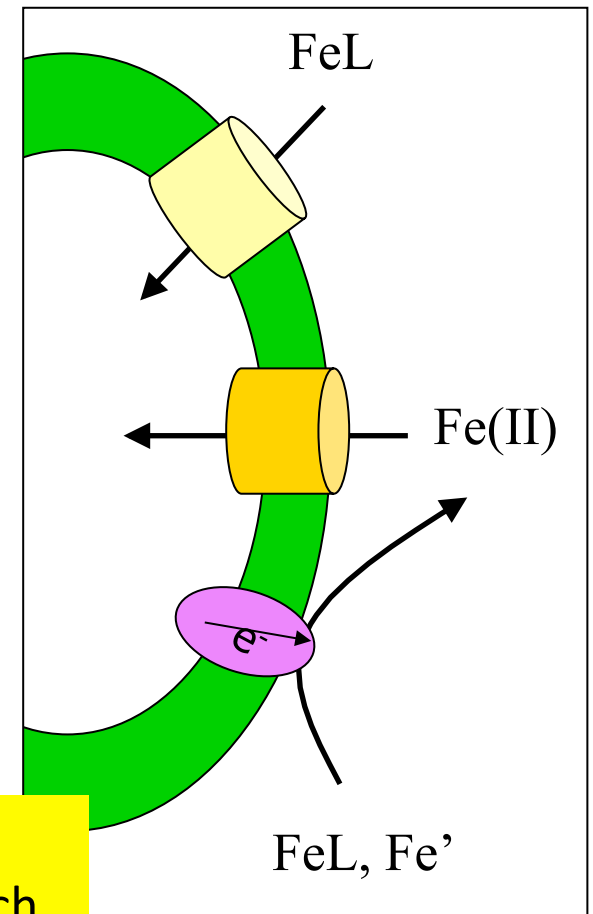
Fundamentals of Fe availability

Emphasis on **kinetics** - Using uptake rate constants (K_{in}) for comparisons & extrapolation to the environment

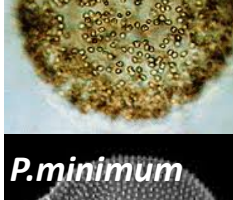
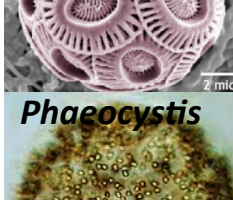
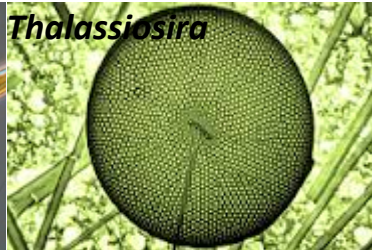
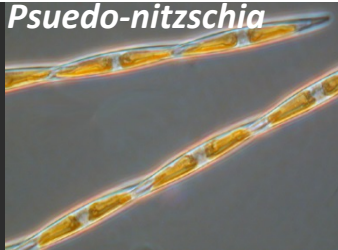
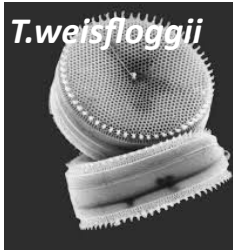
Chemical speciation and kinetics



Uptake pathways & rates

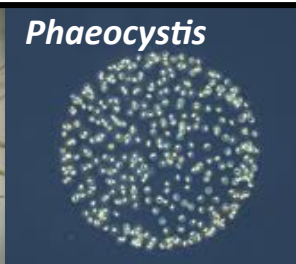
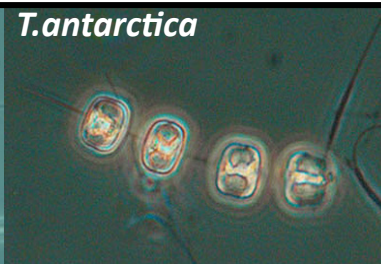
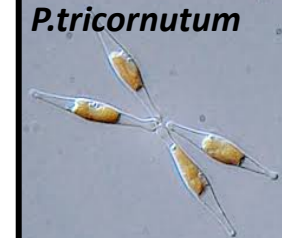
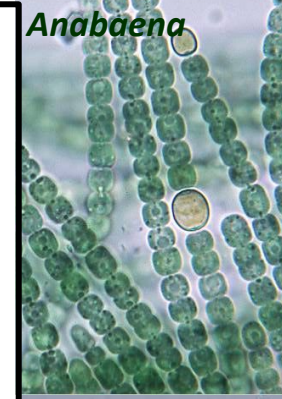


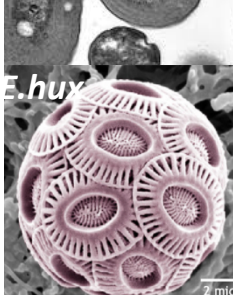
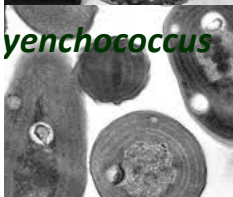
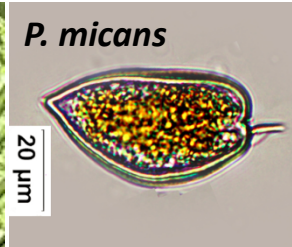
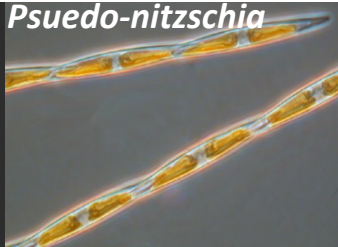
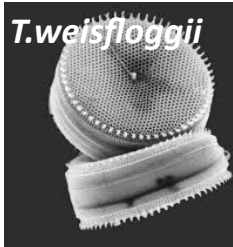
Lis, Shaked, Kranzler, Keren and Morel. 2015. ISME
Iron bioavailability to phytoplankton: an empirical approach



Compiling 5 decades of uptake studies:

- Do phytoplankton differ in their ability to acquire Fe?
- Are there lower/upper limits to uptake rates?
- Which Fe complexes are more/less available?
- Can lab studies help define Fe availability in natural environments?





Compiling 5 decades of uptake studies:

18 studies from **13** research groups

Short term and long term (growth) uptake

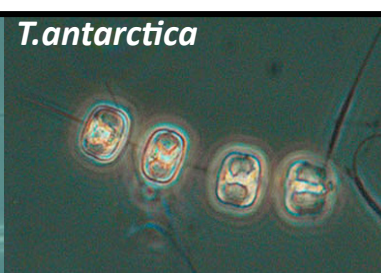
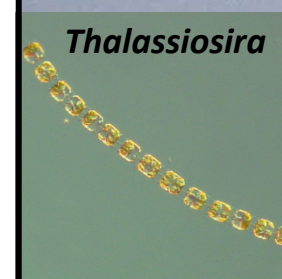
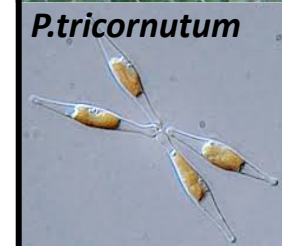
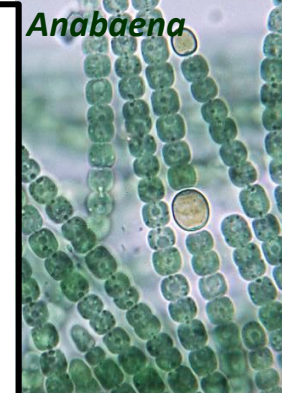
15 phytoplankton species and **28** strains

5 major divisions (Euks & Cyanos)

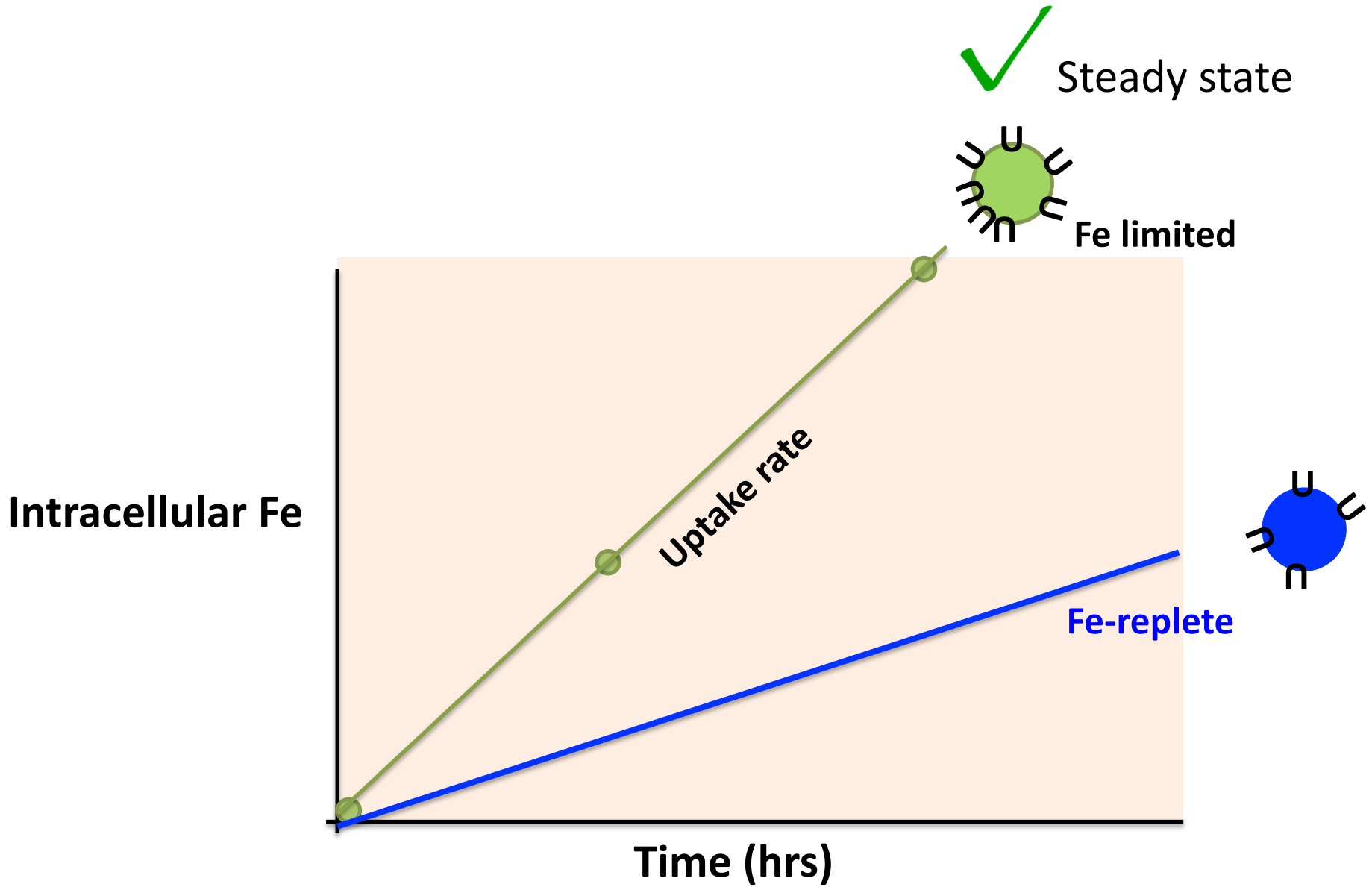
16 Fe-substrates

Stringent data selection criteria
(Fe limited cells, log phase, [Fe] below that of V_{max} , etc..)

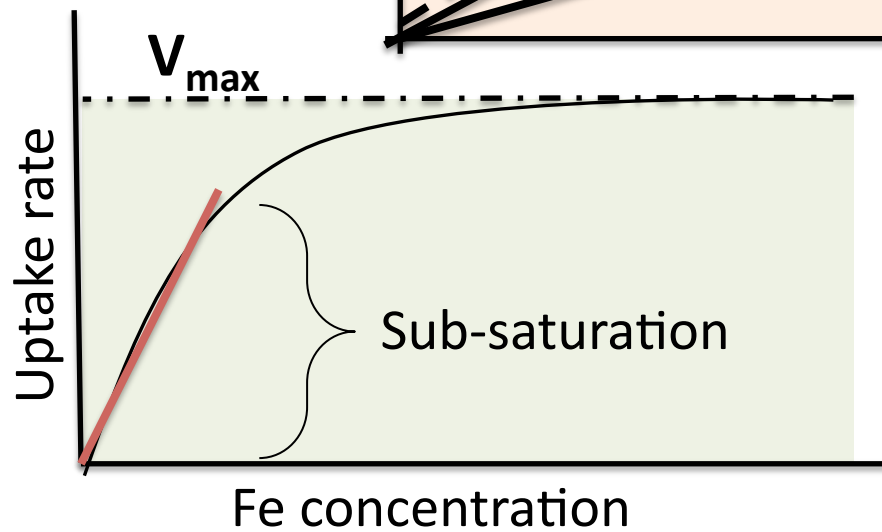
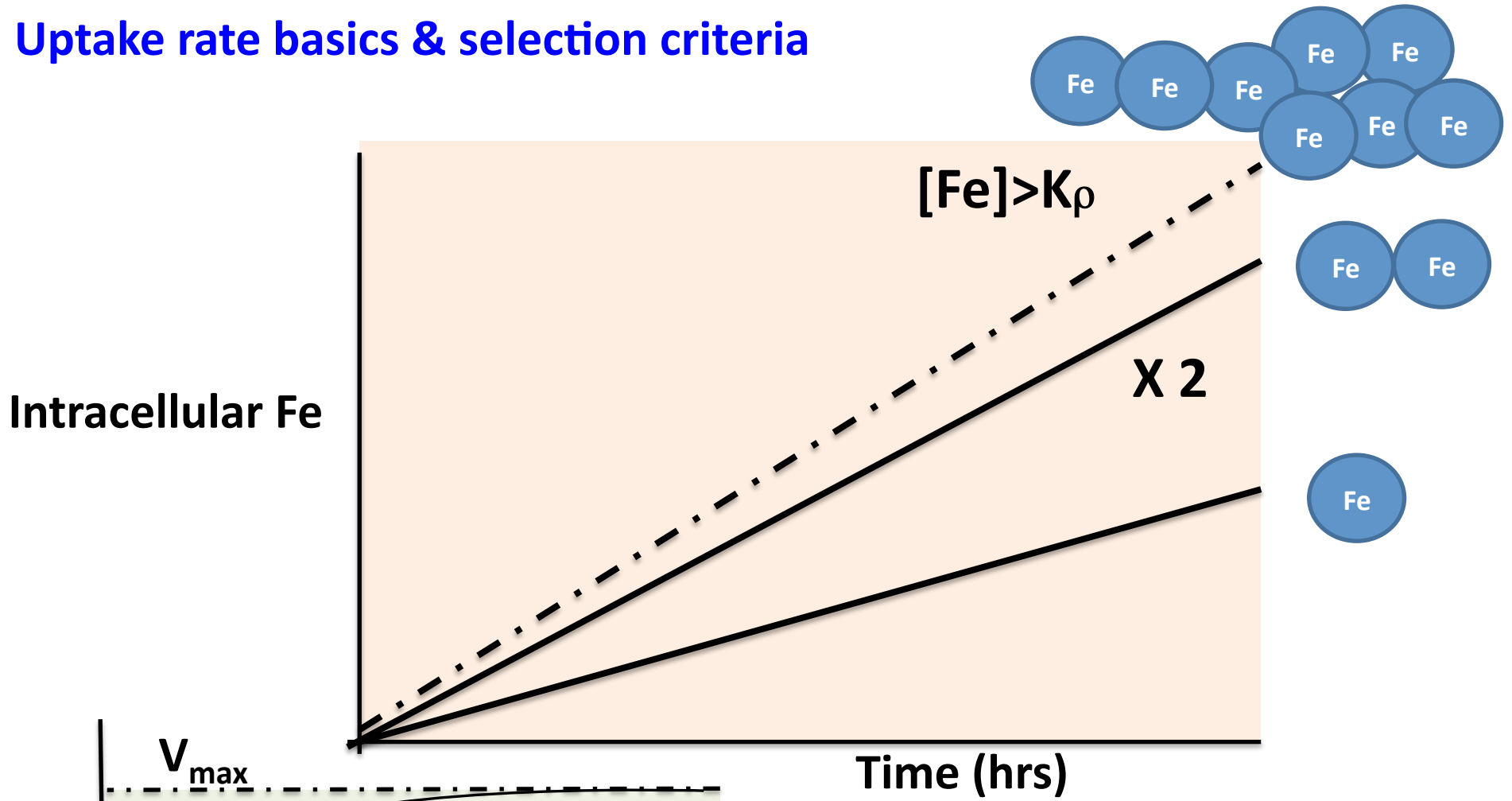
~ 25% our own data



Uptake rate basics & selection criteria

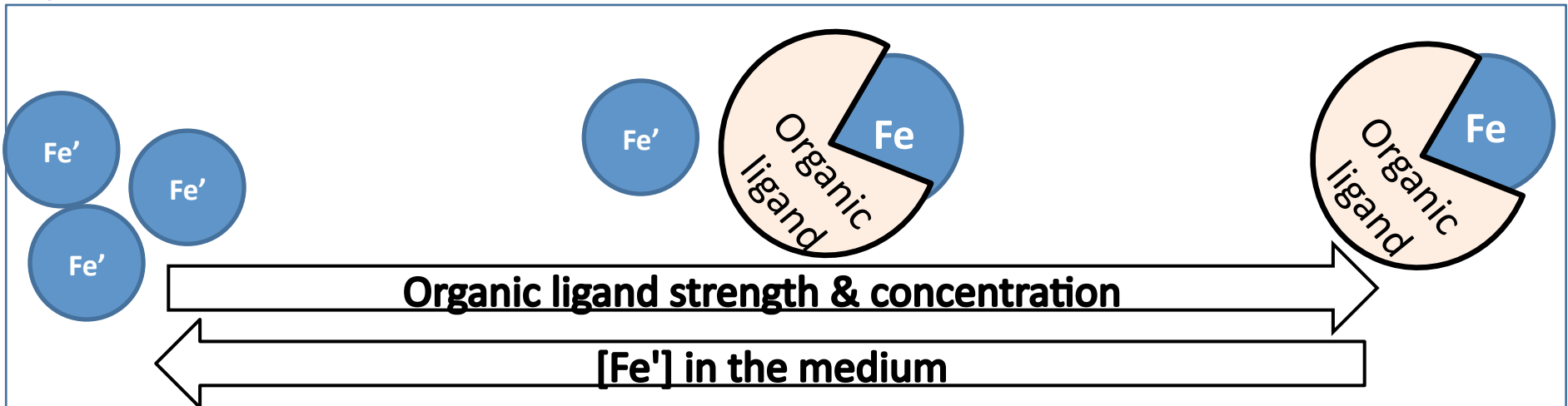


Uptake rate basics & selection criteria

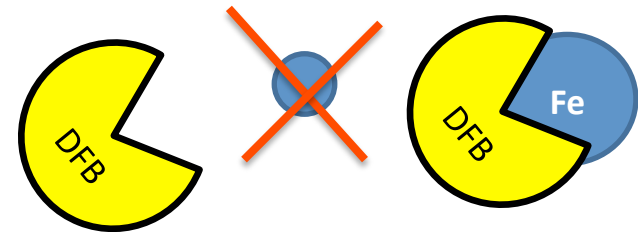


$$\rho = k_{in} \cdot [S]$$

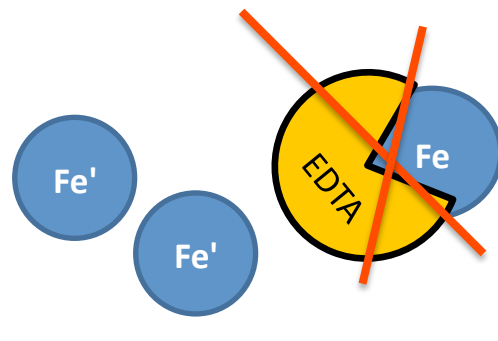
Uptake rate basics & selection criteria



Strong Ligands – FeDFB



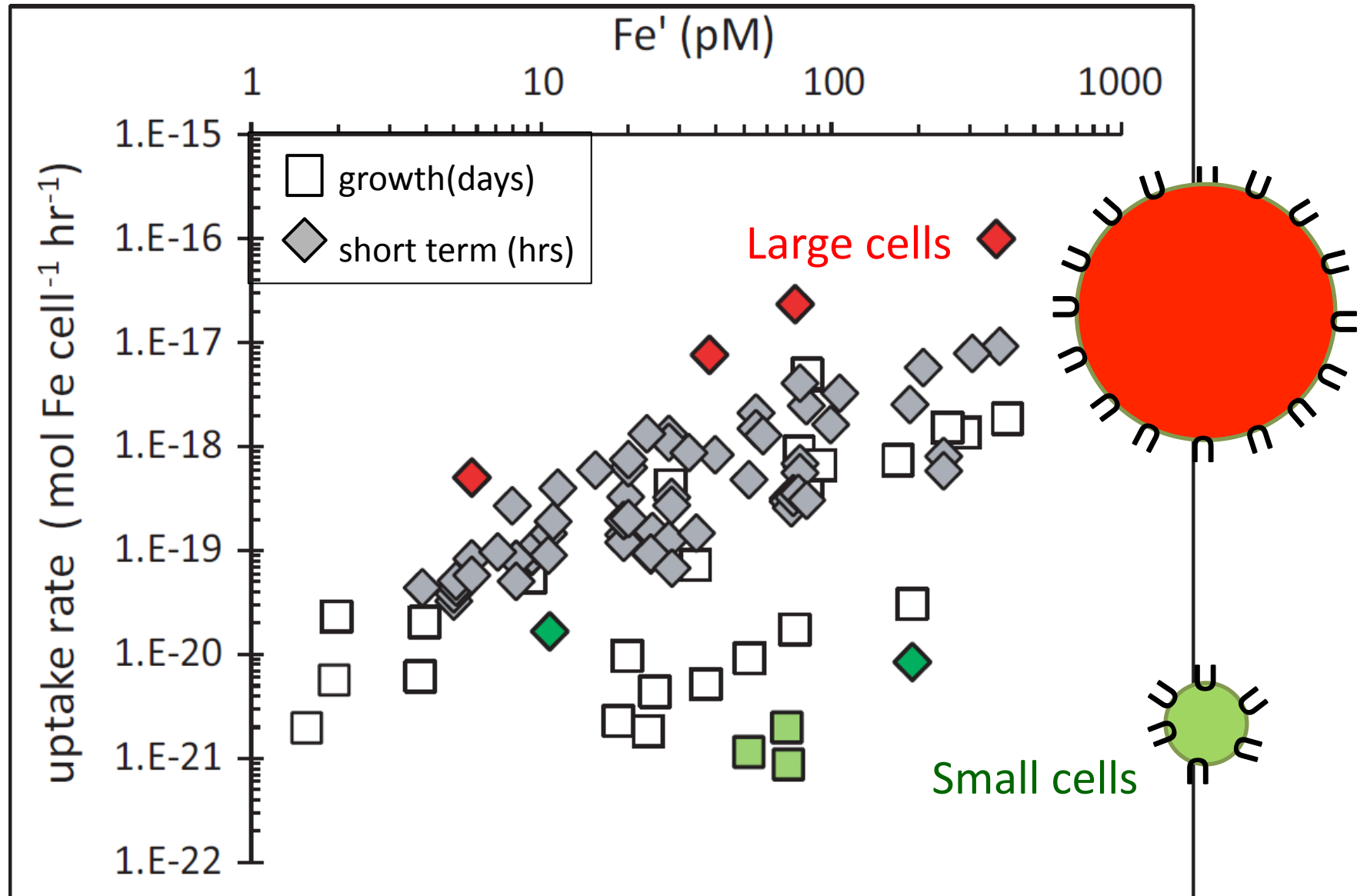
Fe' Fe' **Inorganic Fe – limited by low solubility**



Inorganic Fe – buffered by EDTA

$$\rho = k_{in} \cdot [S]$$

Uptake of inorganic Iron (Fe')

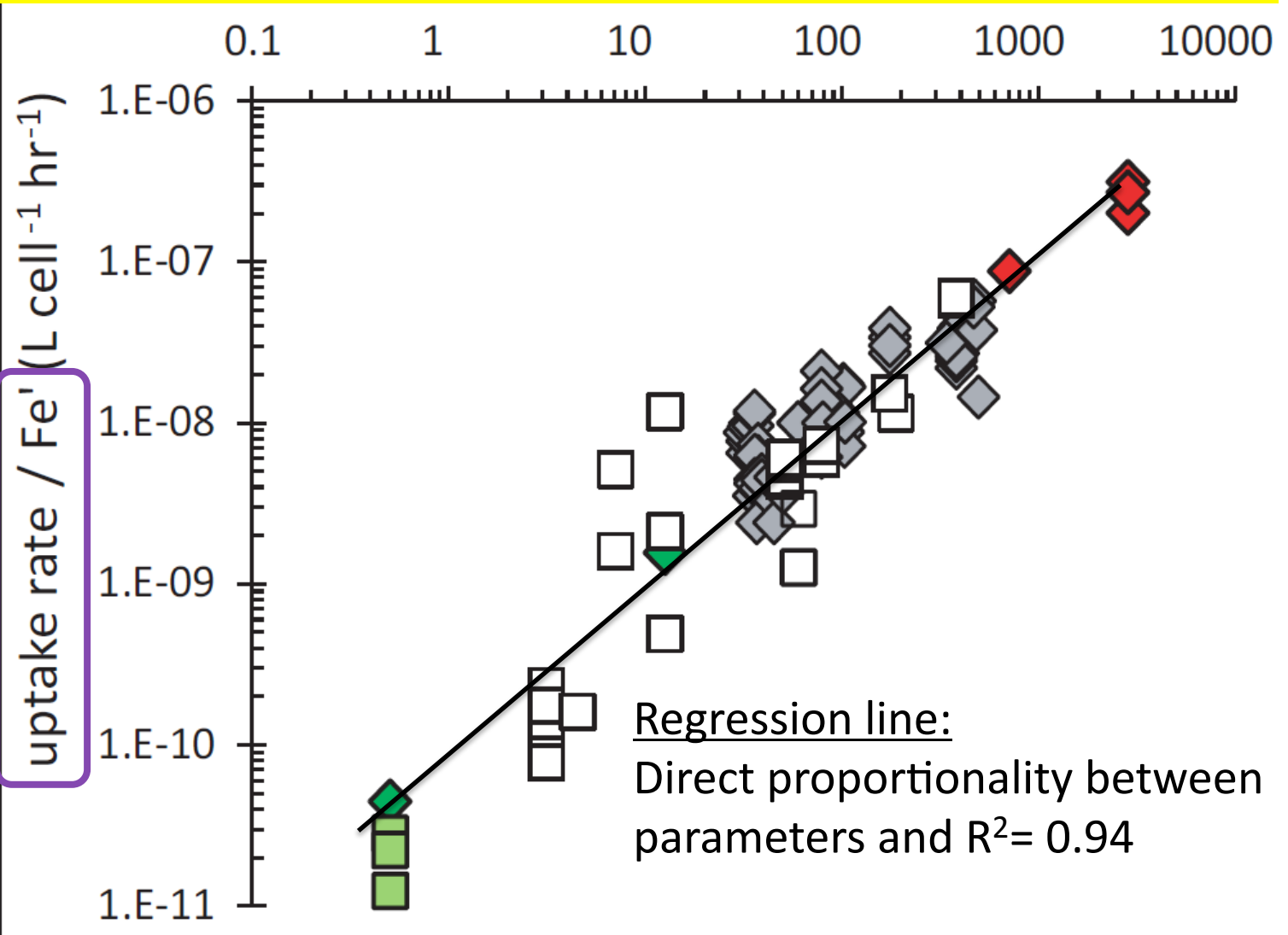


Surface area normalized uptake ($k_{in}/S.A$) is similar among all studied eukaryotes

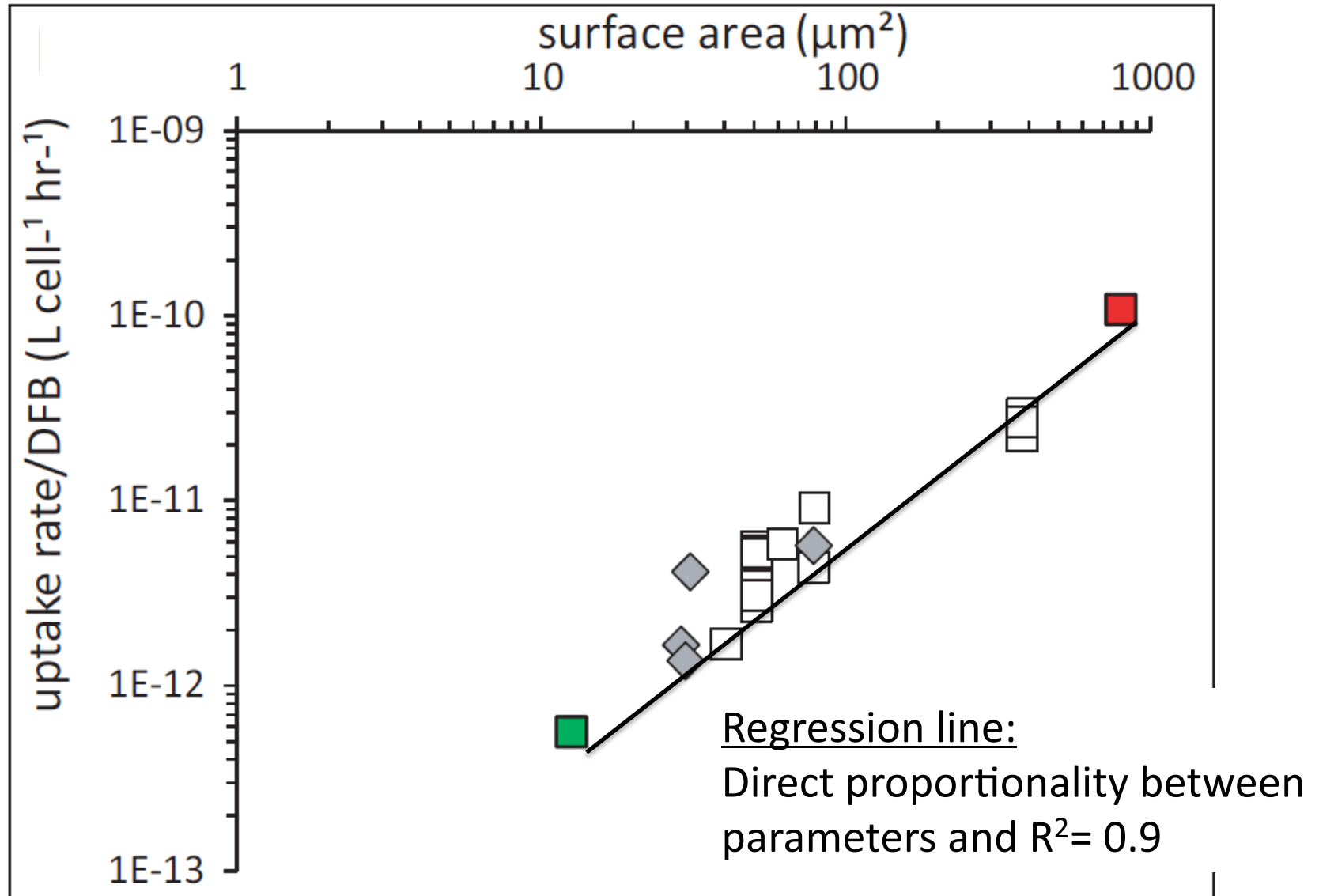
$$\rho = k_{in} \cdot [S]$$



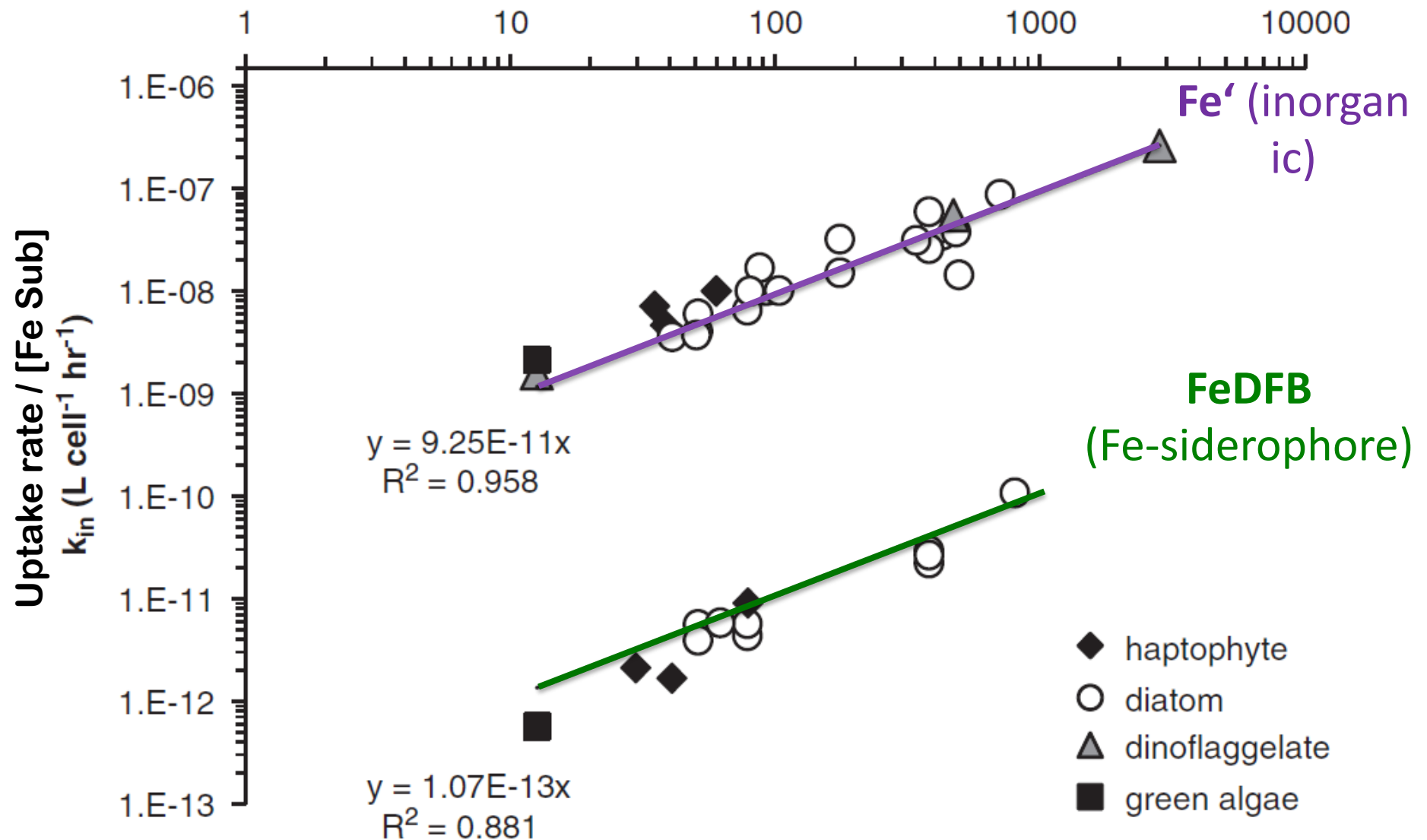
$$k_{in} = \frac{\rho}{[S]}$$



Uptake of siderophore bound Fe (FeDFB)

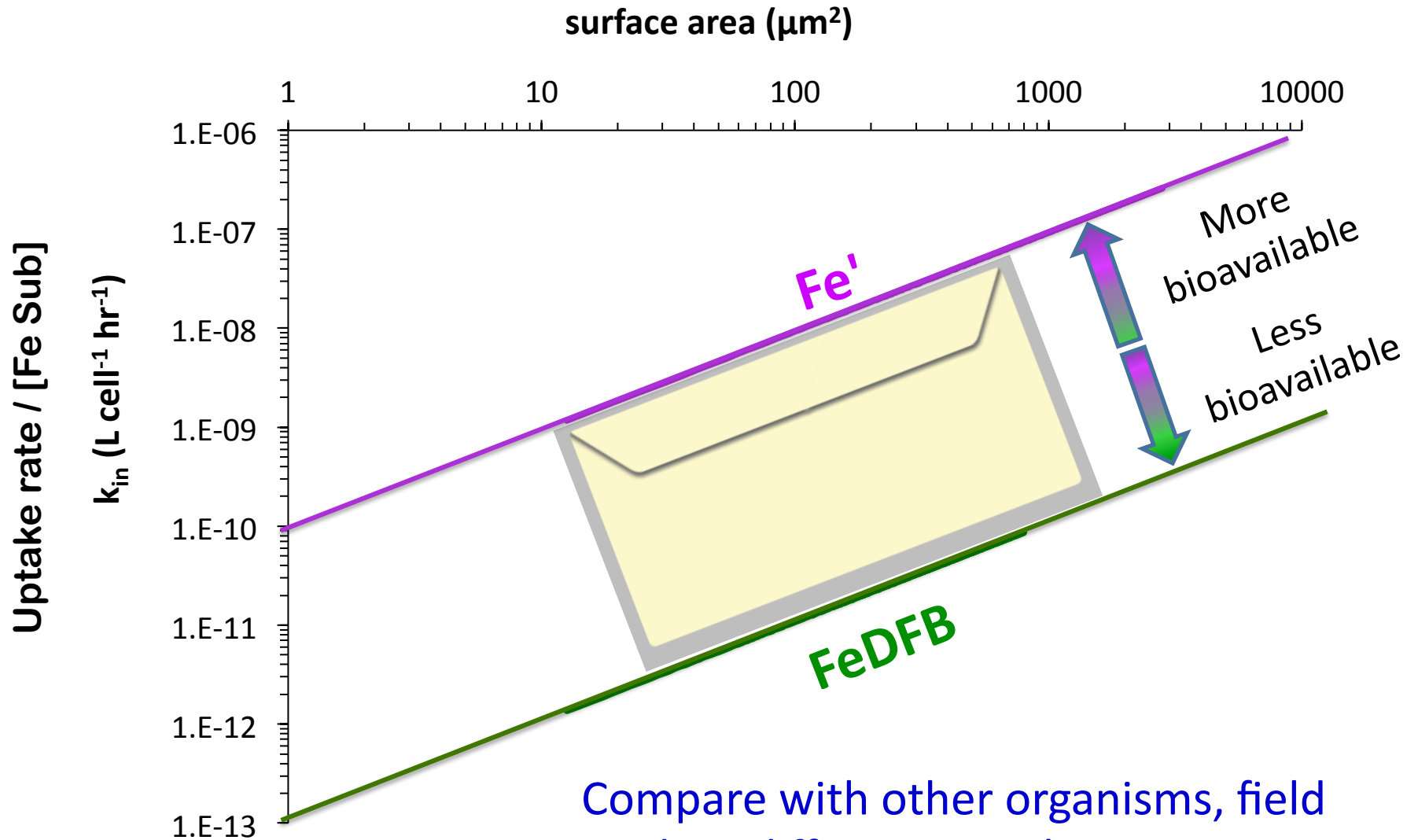


Surface area normalized uptake ($K_{in}/S.A$) is similar among all Euks for Fe' and for FeDFB. But Fe' >> FeDFB (x1000)



The bioavailability envelope

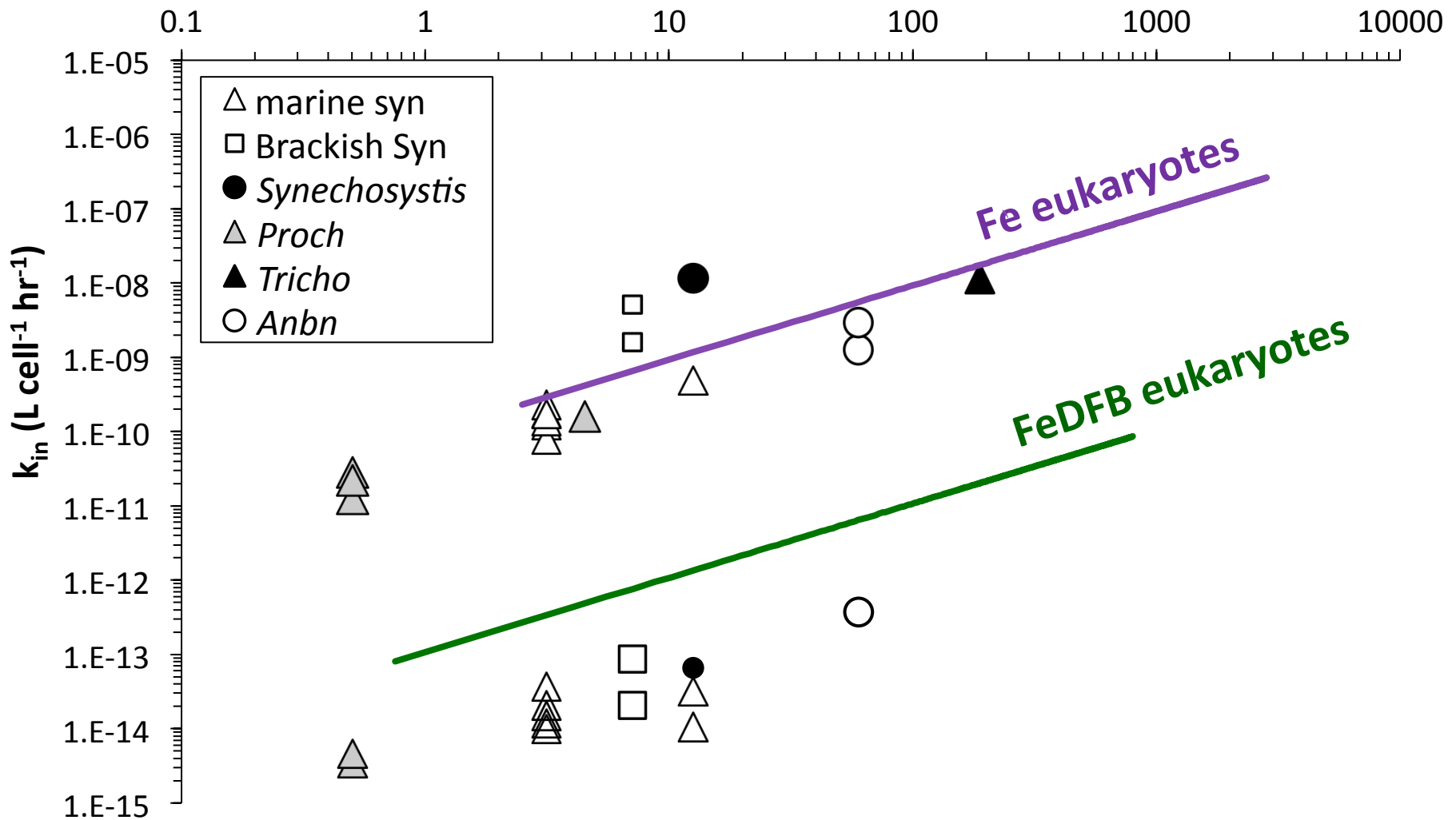
Empirical results of lab studies with euks



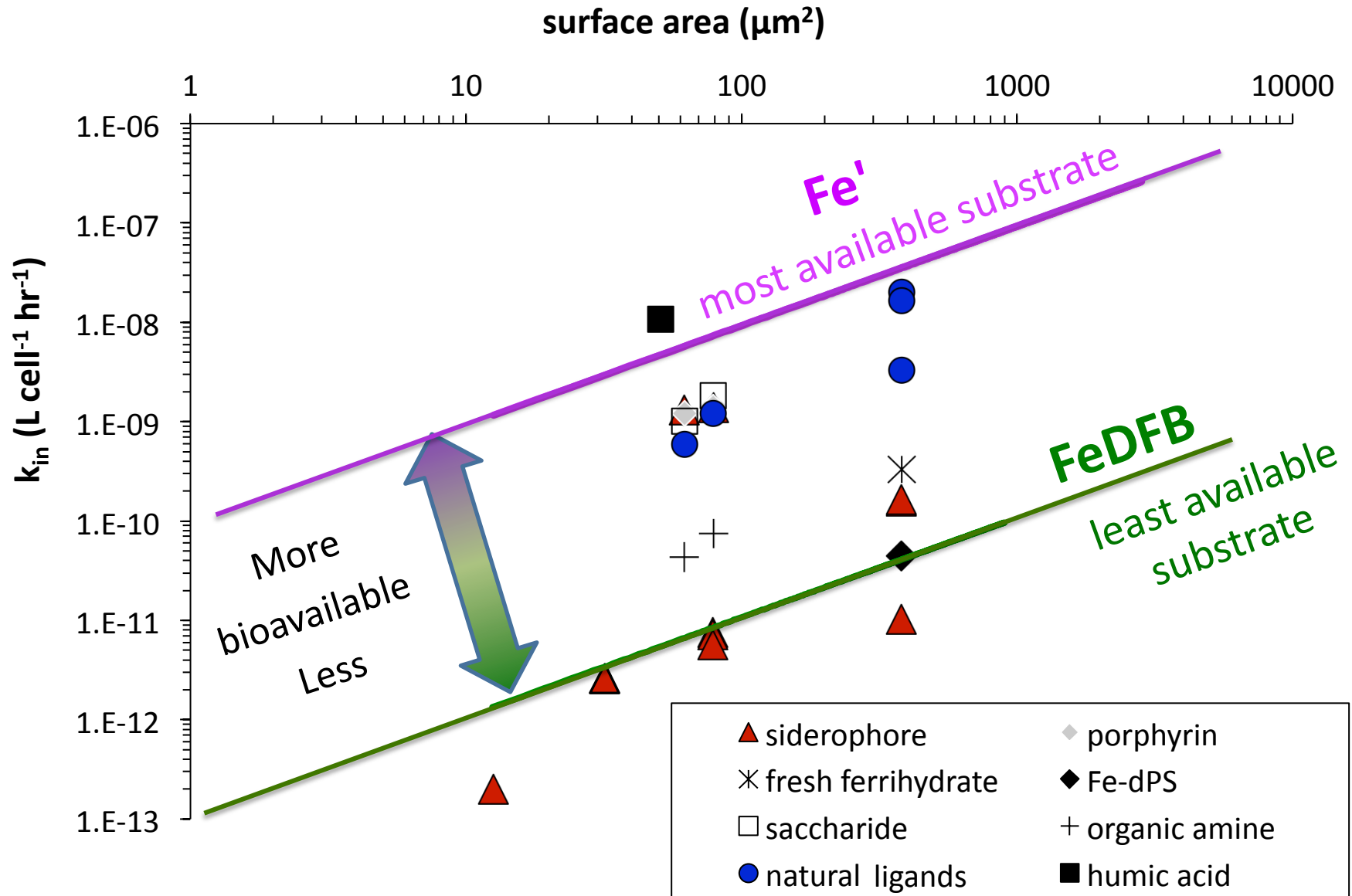
Compare with other organisms, field studies, different Fe substrates...

$k_{in}/S.A$ is similar among all studied organisms for Fe'

FeDFB uptake of cyanos is slower than in euks



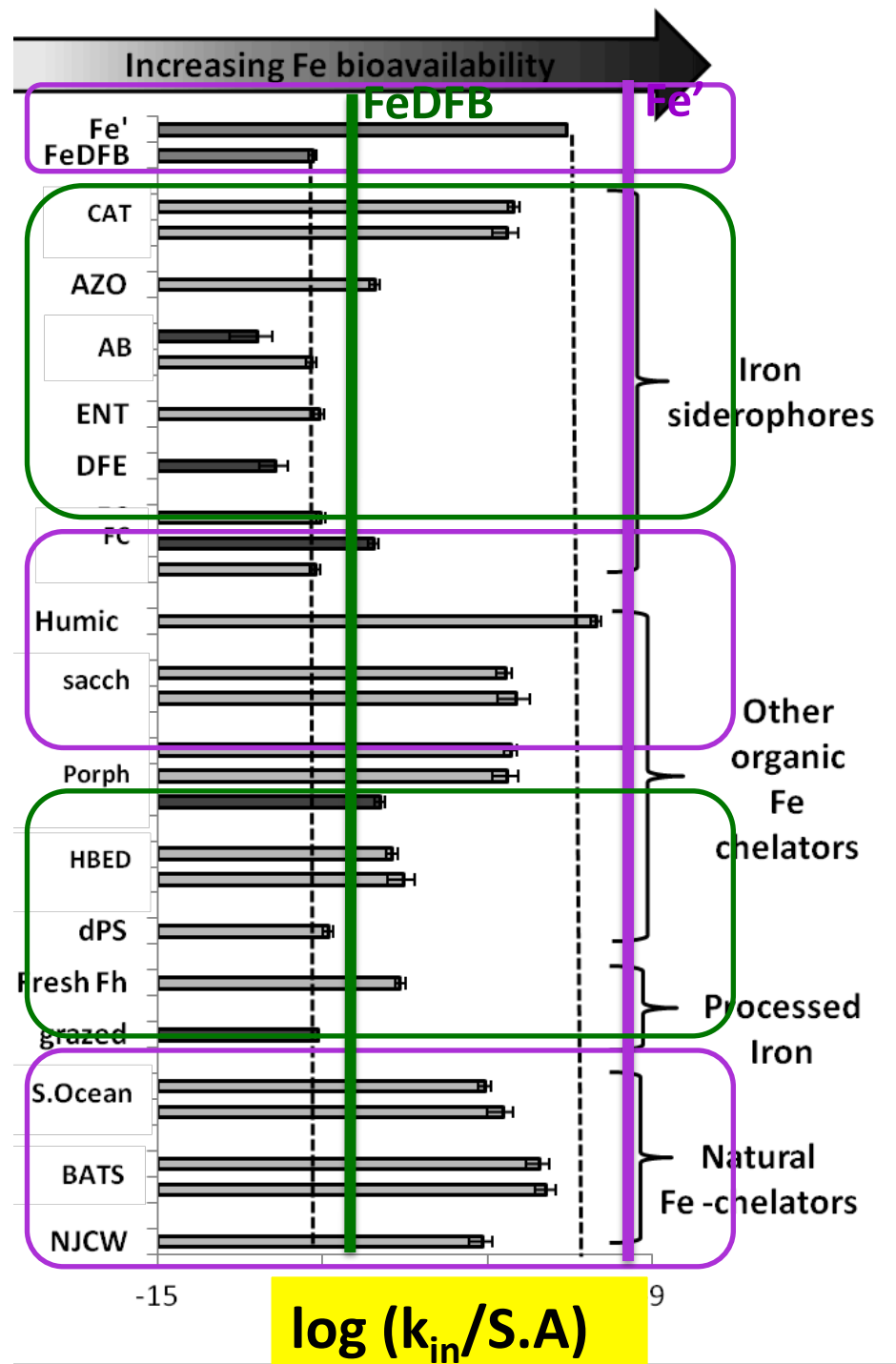
The bioavailability envelope: comparing Fe substrates



The bioavailability envelope: comparing Fe substrates

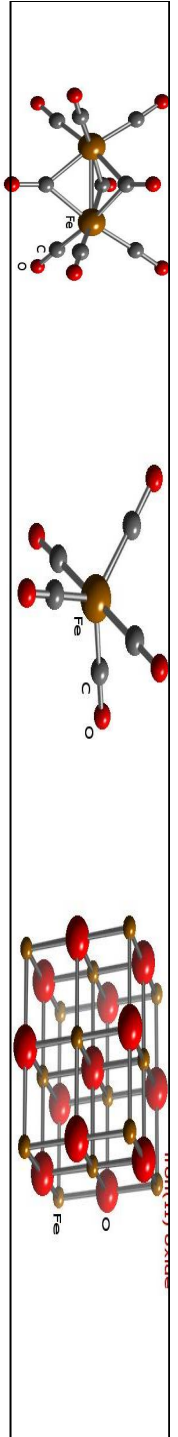
More bioavailable Fe substrates

Less bioavailable Fe substrates



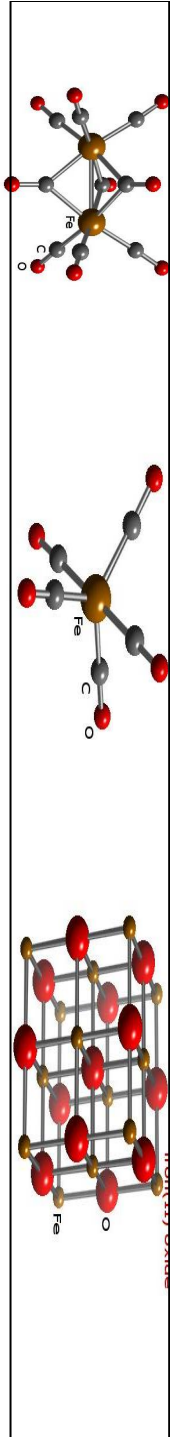
Conclusions and Implications

- Fe uptake rates of defined substrates can be predicted according to cell size.
- We present a convenient framework for addressing bioavailability by comparing across Fe-substrates and organisms.



Conclusions and Implications

- ❖ All phytoplankton are limited by the same fundamental physical, chemical or biochemical factors. Uptake systems have evolved to operate at their maximal efficiency.
- ❖ Phytoplankton may employ a similar iron uptake mechanism : reductive Fe uptake
Experimental data for ~10 cyanos and ~30 euks!!
- ❖ If phytoplankton cannot further increase uptake rates, a competitive advantage in Fe-limited waters must be gained through alternative means.
Decrease cell size, decrease Fe demands, use alternative Fe sources



Availability of Solid Phase Fe to Phytoplankton: Defining major factors at play

Which physico-chemical form can the cell access?

Dissolved

(generated from the particles)

Colloidal/ Particulate

(directly ingested)

The whole pFe pool is available

How soluble is the mineral? (chemical reactivity, [ligand], light etc...)

Availability extracted from solubility & dissolution rates

Can cells enhance mineral dissolution rate?

Add bio-mediated dissolution

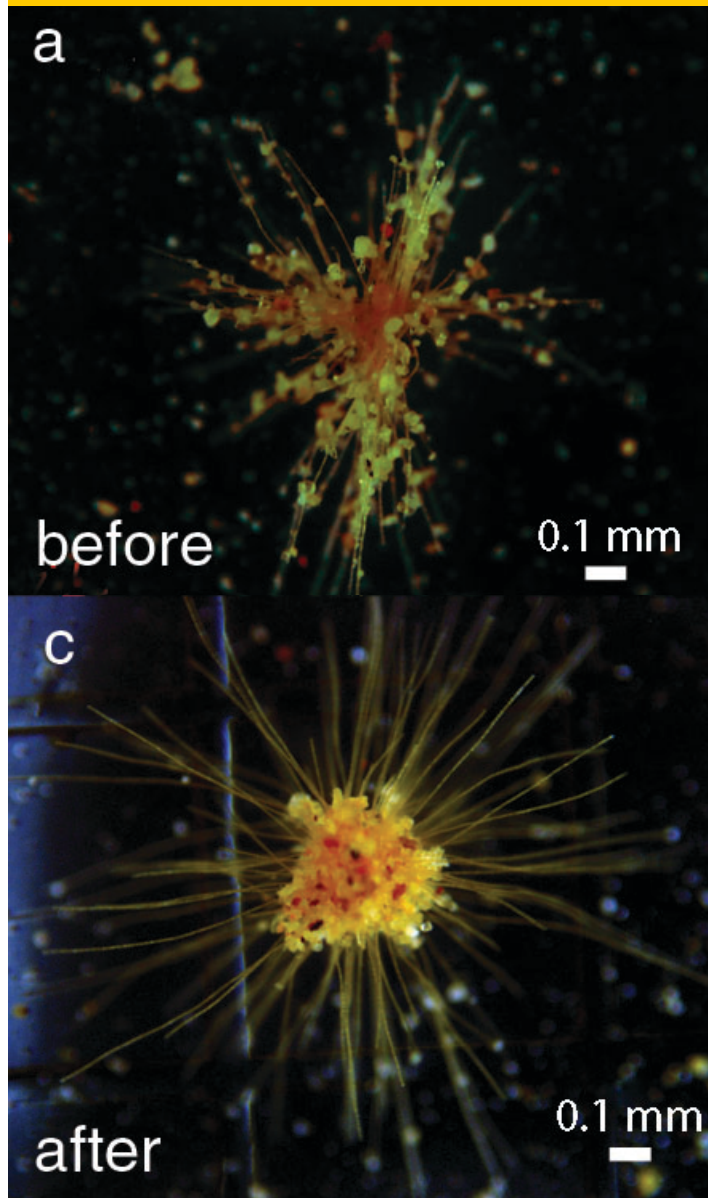
- Removing Fe' (shifting equilibrium)
- Active dissolution by phytoplankton
- Team-up with bacteria (siderophores)

Mineral residence time in upper water

Longer - availability ↑

Can phytoplankton keep mineral Fe at the ocean surface?

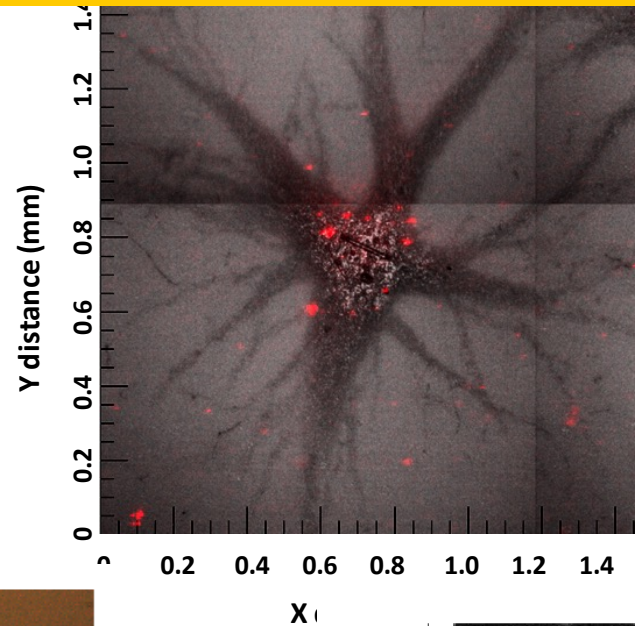
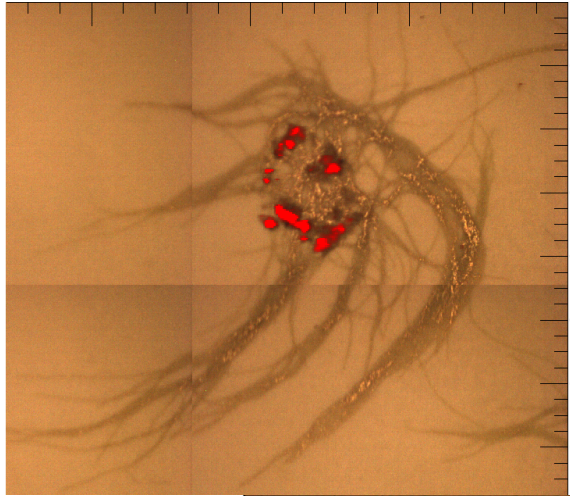
Dust-Fe capturing and modification by natural *Trichodesmium*



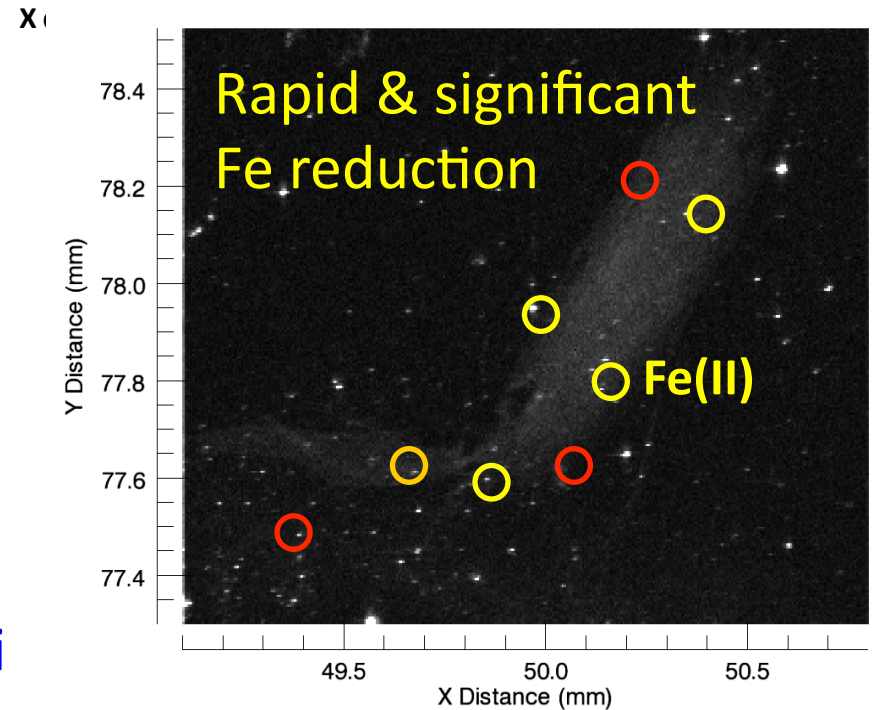
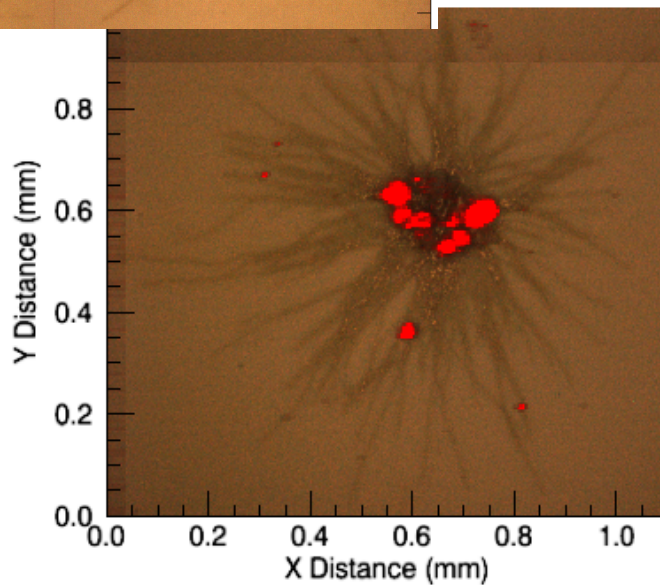
Rubin et al. 2011

Dust-Fe capturing and modifications by natural *Trichodesmium*

Centering of Fe-rich dust

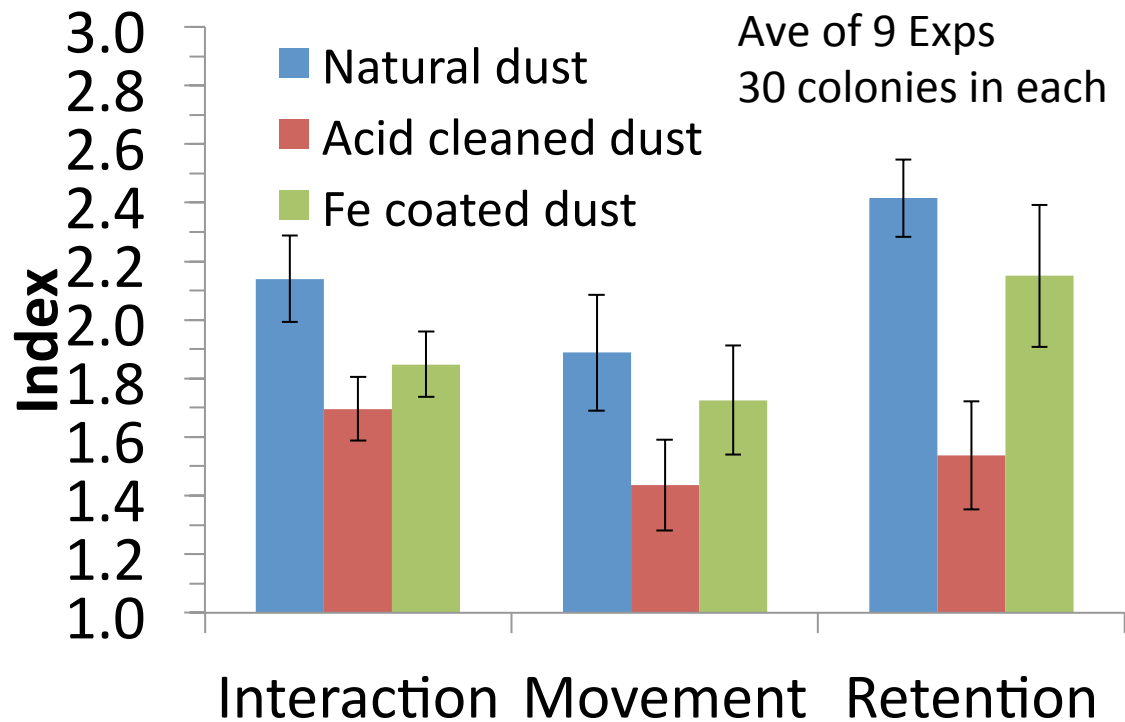
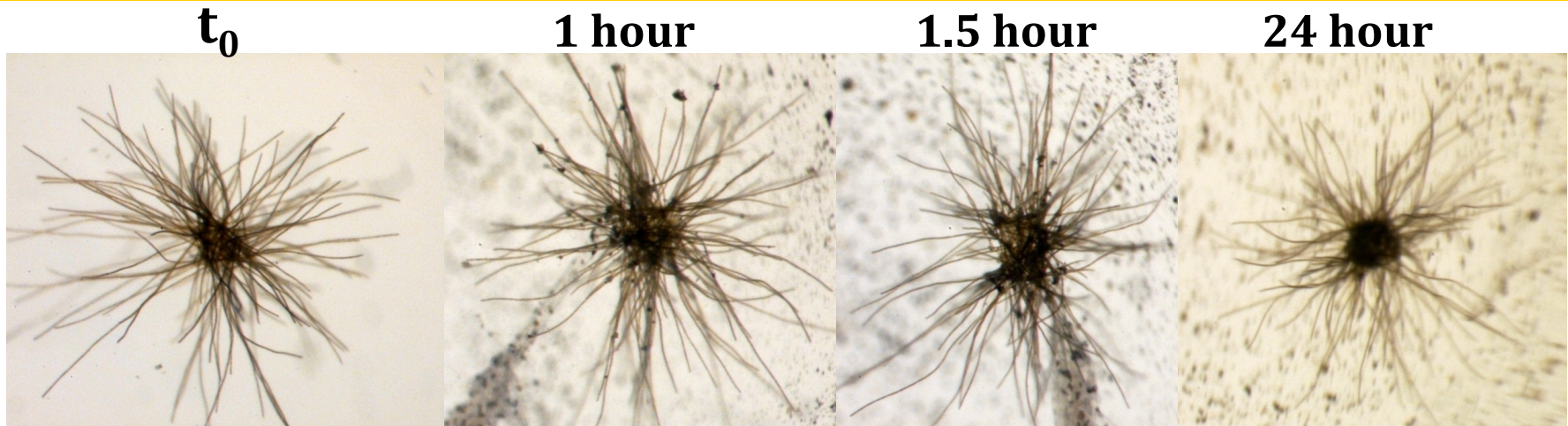


In situ association with Fe-rich particles

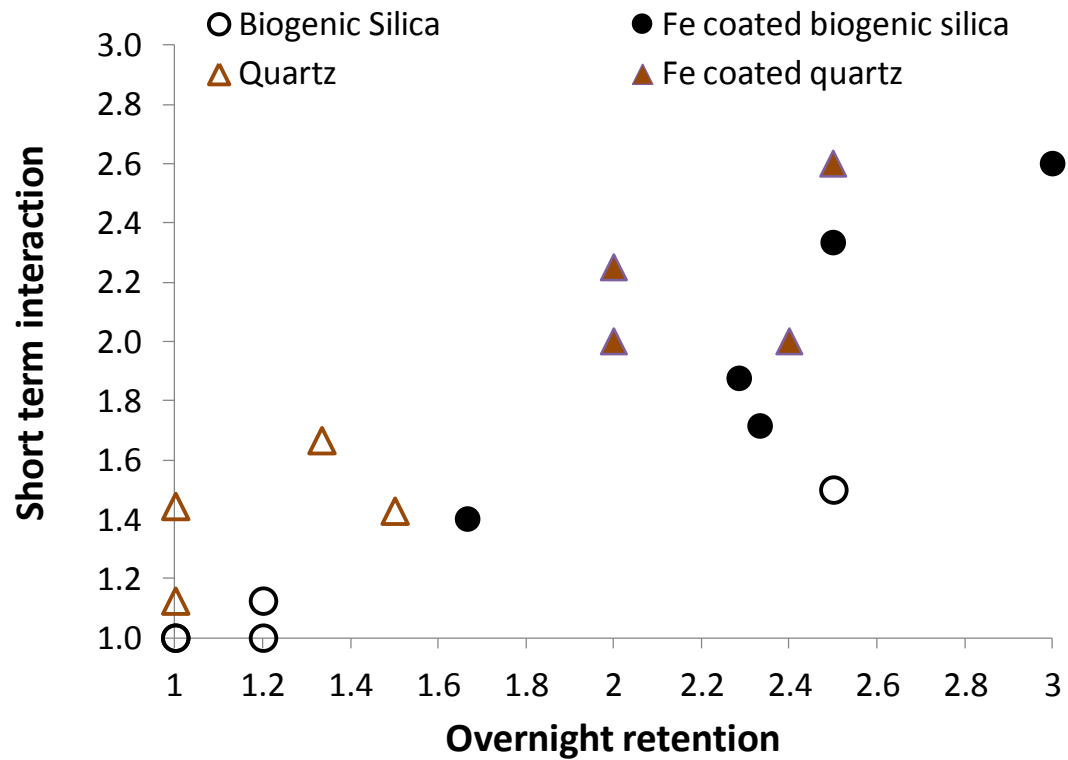
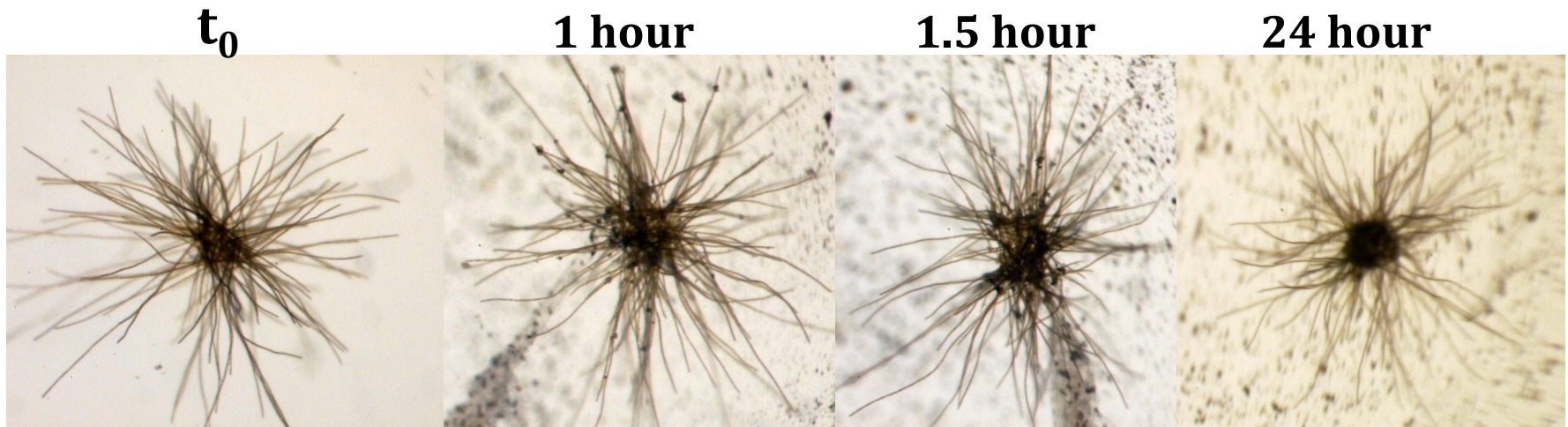


Synchrotron study with Satish Myneni

Sensing of Fe in particles by *Trichodesmium*



Sensing of Fe in particles by *Trichodesmium*

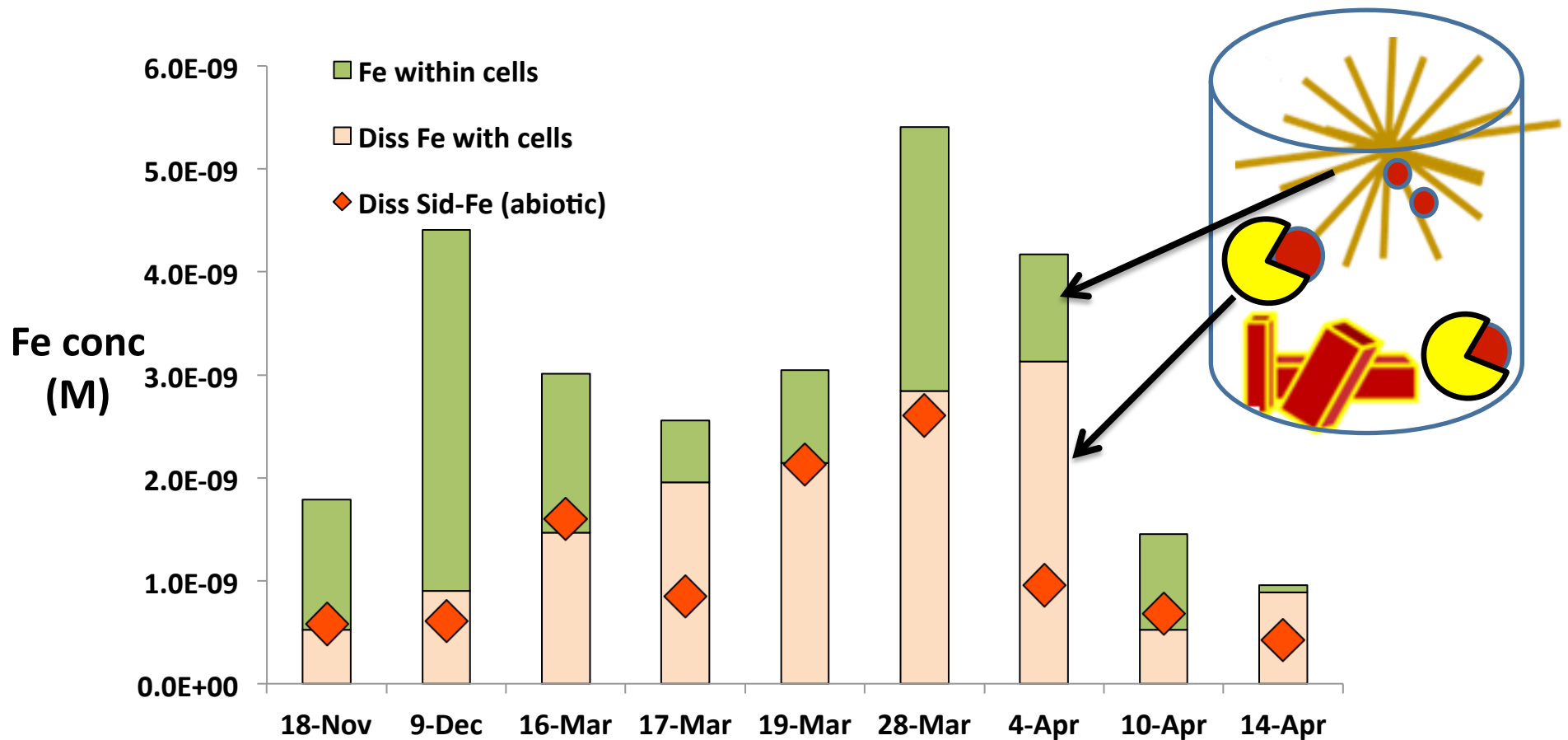


Dust-Fe capturing and **modifications** by natural *Trichodesmium*

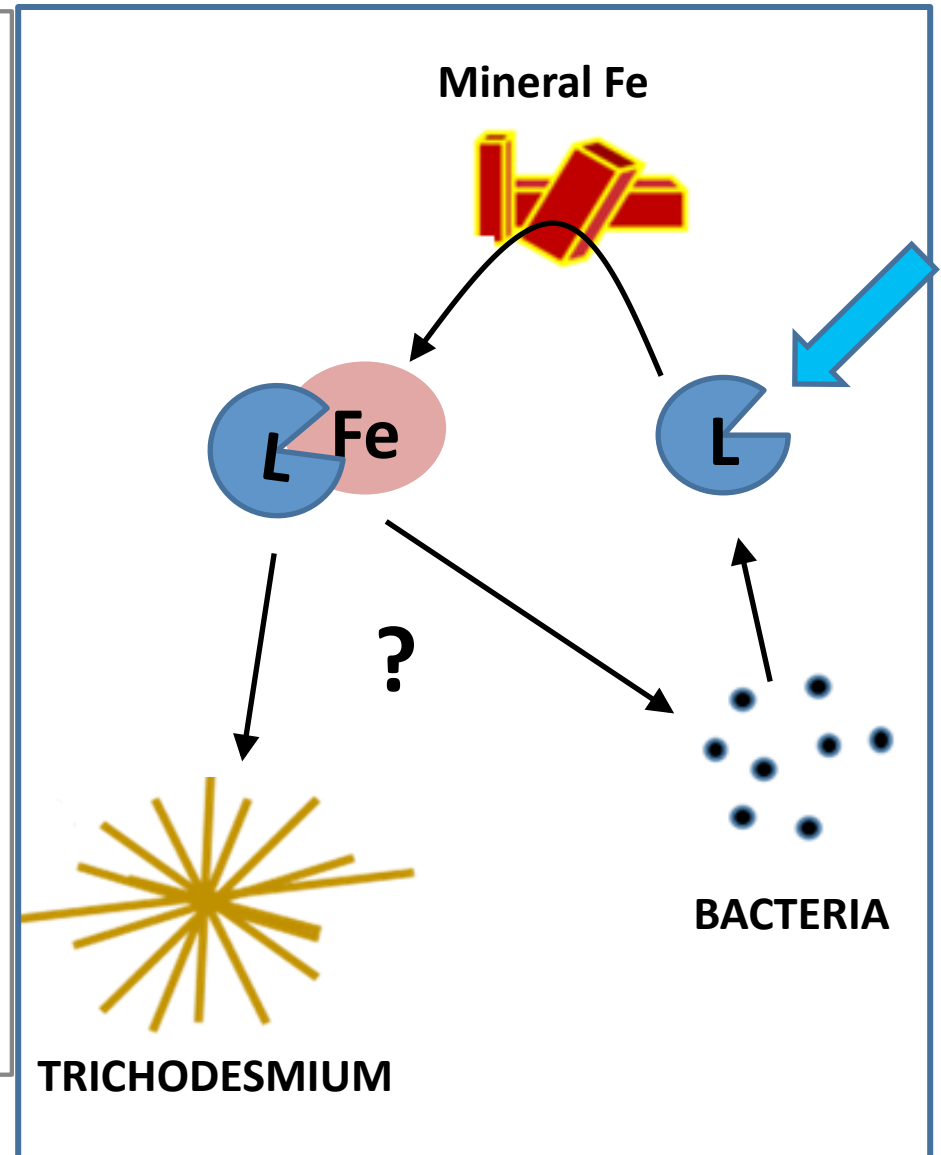
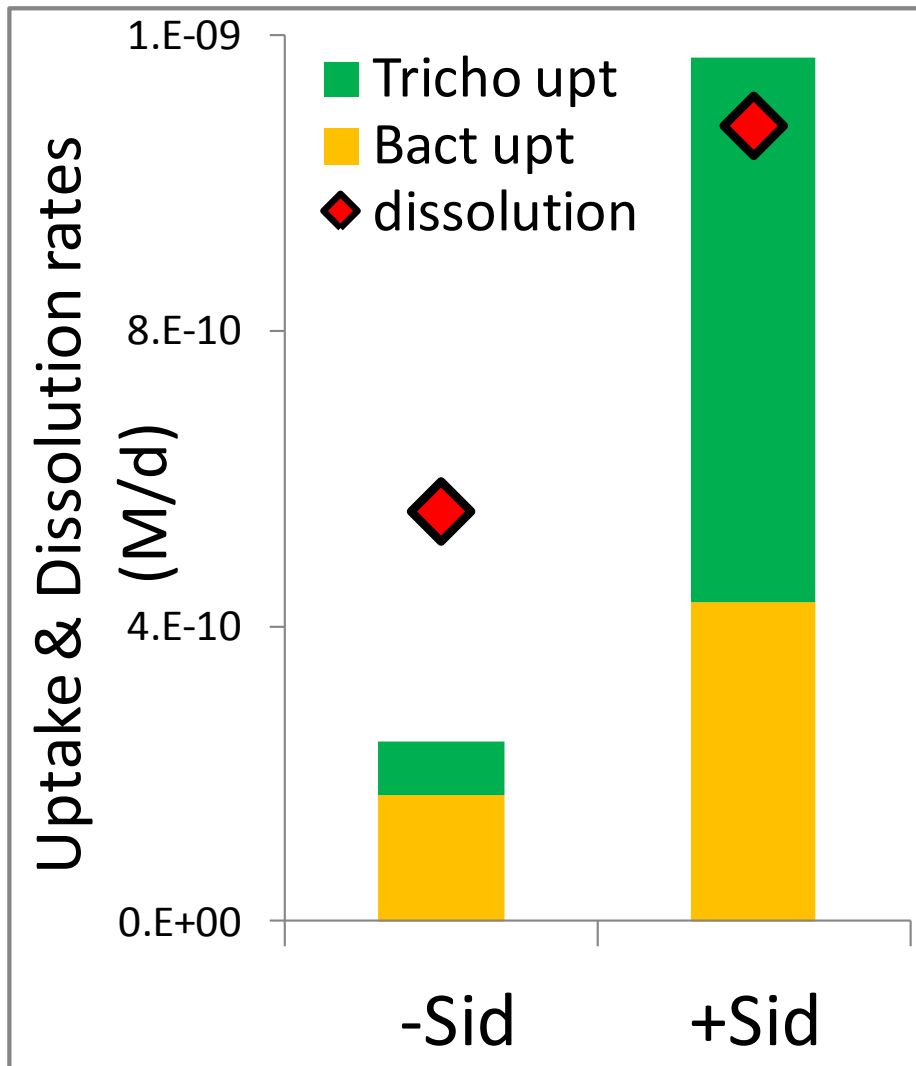
Low level assay for ^{55}Fe -ox dissolution with Sid

Dust-Fe capturing and **modifications** by natural *Trichodesmium*

Bio-mediated ^{55}Fe -ox dissolution

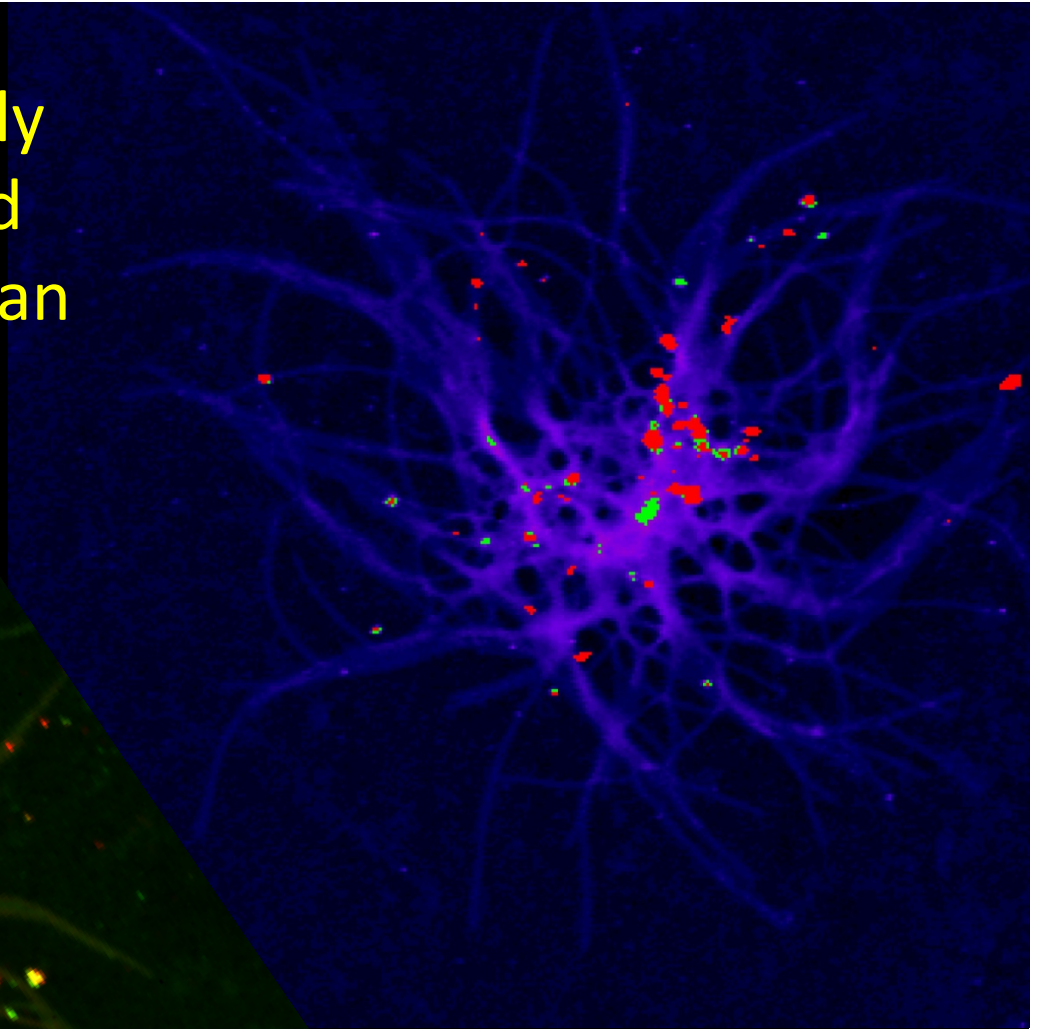
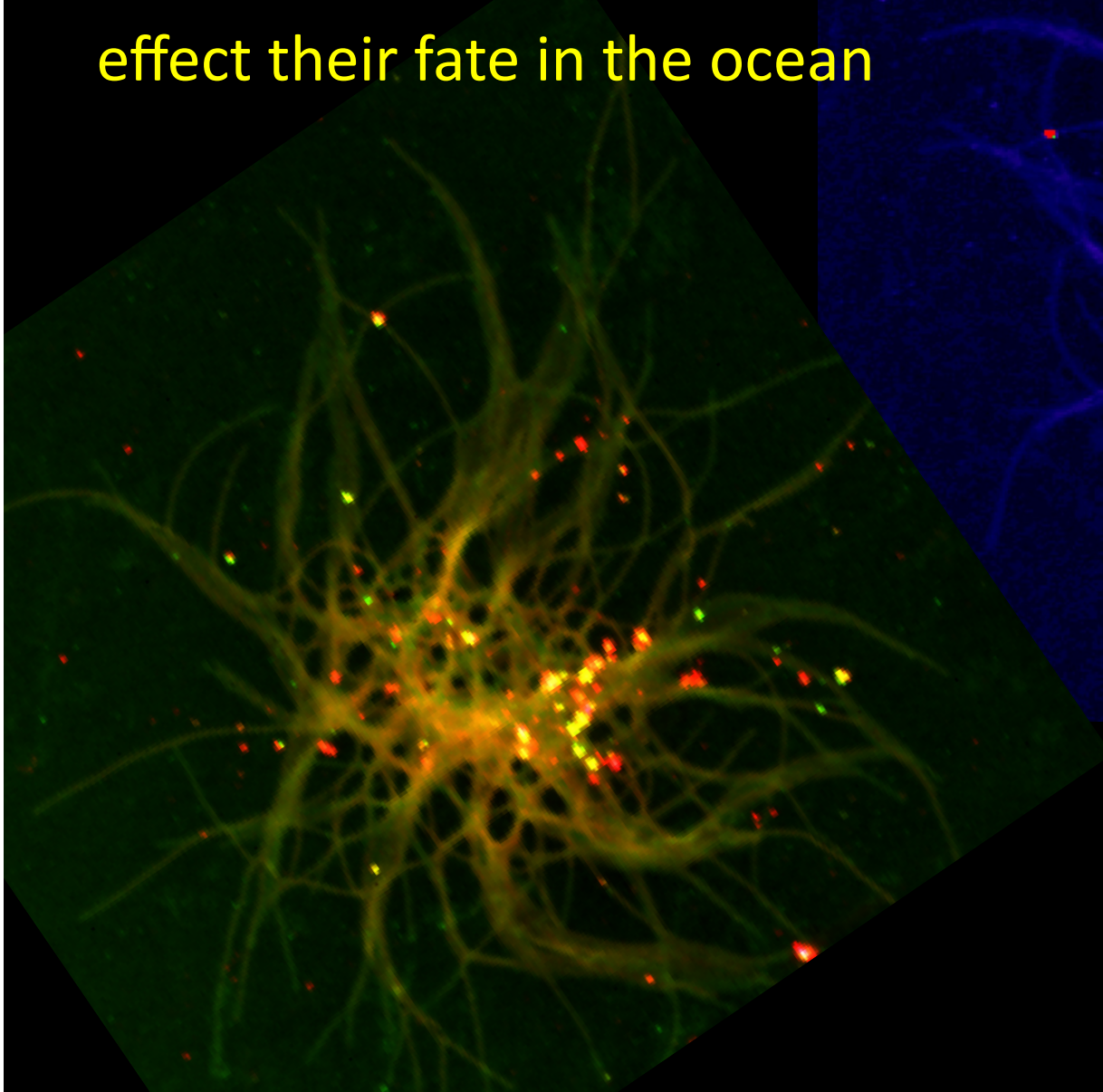


Dust-Fe uptake by natural *Trichodesmium* Assisted by Bacteria



⁵⁵Fe-Ox, Natural colonies, Red Sea, Mar 2016

Phytoplankton may actively mine Fe from minerals and effect their fate in the ocean



Thank you